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Chapter 8 - Chemical
Equations and
Reactions 8-1

Describing Chemical
Reactions I.

Introduction A.

Reactants 1. Original
substances entering
into a chemical rxn B.

Products 1. The
resulting substances
from a chemical rxn

Reactants \rightleftharpoons Products

C. Chemical Equation

1. Represents with
symbols and formulas,

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Chemical

the identities and
relative amounts of the

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**Chapter 8 - Chemical
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Equations and
Reactions. STUDY.

Flashcards. Learn.

Write. Spell. Test.

PLAY. Match. Gravity.

Created by.

Bavly_Halaka. Terms in
this set (46) Proof 2

that a chemical

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reaction has occurred:

Production of a gas.

The evolution of gas bubbles when two

substances are mixed

is often evidence of a chemical reaction. For

example ...

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Flashcards ...

Chemical Equations

and Reactions The

evolution of energy as

light and heat is an

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indication that a
chemical reaction is
taking place. CHAPTER
8 Fireworks. CHEMICAL
EQUATIONS AND
REACTIONS 261
SECTION 1 O

OBJECTIVES List three
observations that
suggest that a
chemical reaction has
taken place.

CHAPTER 8 Chemical Equations and Reactions

Chapter 8: Chemical

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Chapter 8

Chemical Equations and

Reactions. I. Describing

Chemical Reactions. A.

A. chemical reaction. is

the process by which

one or more

substances are

changed into one or

more different

substances. A.

chemical. equation.

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Equations and

Reactions

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equations reactions

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Cumbustion reaction
Practice balancing
equations. Types of
chemical reactions.
What type of reaction
is it? Practice balancing

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equations. The

reactivity series of

metals . Chapter labs:

Evidence for chemical

change Chapter

homework: Section 1:

Chapter review 1 thru

10. Section 2: Chapter

review 18 thru 21.

Section 3: Chapter

review 31 thru 33 ...

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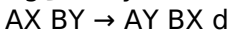
CHAPTER 8 REVIEW
Chemical Equations
and Reactions SECTION
2 SHORT ANSWER
Answer the following
questions in the space

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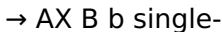
Chapter 8

Chemical

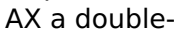
provided. 1. Match the equation type on the left to its representation on the right. c synthesis (a)



d decomposition (b) $A + BX$



b single-displacement (c) $A + B \rightarrow$



a double-displacement (d) $AX \rightarrow$



8 Chemical Equations and Reactions

What type of chemical

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reaction does the following chemical equation represent?

$2\text{HCl}(\text{aq}) + \text{Cr}(\text{s})$ yields to $\text{H}_2(\text{g}) + \text{CrCl}_2(\text{aq})$...

Chapter 8. OTHER SETS BY THIS CREATOR. 20

terms. Chapter 6

vocabulary anatomy.

64 terms. The Fall of the Roman Empire and the Rise of the Imperial Church.

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8-2 Balancing Chemical

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Chemical Equations Atoms
rearrange in chemical
reactions, but atoms
cannot be created or
destroyed (Law of
Conservation of
Matter). For this
reason, the number of
each kind of atom on...

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...

1. To balance a
chemical equation, it is
permissible to adjust
the a. coefficients. b.
subscripts. c. formulas

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Equations
Reactions Answer Key

of the products. d.
number of products., 2.

In a chemical equation,
the symbol (aq)

indicates that the

substance is a. water.

b. dissolved in water. c.

an acid. d. insoluble.,

3. The production of a

slightly soluble solid

compound in a double-

displacement reaction

results in the ...

Chapter 8: Chemical Equations and Reactions

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Chapter 8

Chapter 8 is the first chapter of the second semester. Here, we will cover chemical reactions and equations. Students will first learn about the basics of chemical reactions and how to recognize...

Chapter 8 - Chemical Equations & Reactions - yazvac

Chapter 8: Chemical Equations and Reactions To balance a

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Chapter 8

Chemical Equations Answer Key

chemical equation, it is permissible to adjust the

1. coefficients.
2. subscripts.
3. formulas of the products.
4. number of products.

., 2. In a chemical equation, the symbol (aq) indicates that the substance is

1. water.
2. dissolved in water.
3. an acid.
4. insoluble.

., 3.

Chapter 8: Chemical Equations and Reactions

Chemical Equations

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**Chemical Equations
And Reactions**

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Worksheet Chapter

8... Equations

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Reactions Chapter 8

Holt Chemistry...

Chemical reactions and
equation is very

fundamental chapter

that deals with basic

knowledge of chemical

reactions and

equations. Students

are advised to get

tuned-in with this

chapter to avoid any

sort of difficulty in

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understanding

advanced topics of

chemistry. NCERT

Solutions Class 10

Science

Chemical Equations

And Reactions

Chapter 8

chemical equation

represents, with

symbols and formulas,

the identities and

relative amounts of

reactants and products

in a chemical reaction.

The relative amounts

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Chemical

of reactants and products represented in the equation must be adjusted so that the numbers and types of atoms are the same on both sides of the equation.

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A chemical equation is balanced so that the numbers of atoms of each type involved in a chemical reaction are

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the same on the reactant and product sides of the equation. A balanced chemical equation satisfies the law of conservation of mass, i.e., the total mass of the reactants is equal to the total mass of the products.

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