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contents: thermodynamics - chapter 01: thermodynamic properties and state of pure substances. chapter 02: work and heat. chapter 03: energy and the first law of thermodynamics. chapter 04: entropy and the second law of thermodynamics. chapter 05: irreversibility and availability

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2 sat. = 1.0 atm. Calculate: 1) The vapor mole fraction of species 1 and the total pressure when two liquid phases and a vapor phase are in equilibrium with a liquid phase containing a mole fraction. x.

Chemical Engineering Thermodynamics II

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2 3 energy J N m kg m power = = = time s s s charge current = time charge = current*time = A s energy power = = current*electric potential time 2 3 energy kg m electrical potential = = current*time A s electrical potential current = resistance 2 23

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Chemical Engineering Thermodynamics

Problem solving approach to solve closed system energy balance. Made by faculty at the University of Colorado Boulder Department of Chemical and Biological Engineering.

Problem Solving Approach

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Solution First we must find the amount of heat released by the ethane. To do this, we calculate the number of moles of ethane gas using the ideal gas equation and multiply the molar heat of combustion by the number of moles.

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