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reacts with chlorine to make hydrogen
chloride. The reaction is exothermic. The
reaction can be represented by the
equation below. $H_2 + Cl_2 \rightarrow 2HCl$
 $\Delta H = -184 \text{ kJ/mol}$ (a) A
mixture of 2.5g of hydrogen and 142g of
chlorine is allowed to react.

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