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A.  $1.20 \times 10^3$  B.  $3.01 \times 10^3$  C.  $6.02 \times 10^3$  D.  $1.20 \times 10^4$

2. 1 mol of a hydrocarbon with general formula  $C_nH_{2n+2}$  reacts completely with oxygen to produce 4 mol of  $H_2O$ . What is the amount of oxygen molecules, in mol, that reacts? A. 4 B. 5 C. 6 D. 7

3. Under which combination of conditions is 1 mol

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of an ideal gas present? Volume  
Pressure ...

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$\times \times - 1 \times 1000$  17. Under which conditions will the reaction between 1.0 g calcium carbonate and excess hydrochloric acid be the fastest? Assume that all reactions are carried out at the same temperature. A. One large piece of calcium carbonate and 2 mol dm<sup>-3</sup> hydrochloric acid B. One large piece of calcium carbonate and 1 mol dm<sup>-3</sup> ...

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