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Handbook Of Chemical Engineering Calculations

PHYSICAL AND CHEMICAL PROPERTIES 1.5 where T c, P c, V c, and Z c are critical temperature, critical pressure, critical volume, and critical compressibility factor, respectively. Thus, for p-xylene, T c = 412.3[0.567+0.106−(0.106)2]−1 = 623.0K(661.8 F)(literature value is 616.2 K) P c = 106.16(0.34+1.378)−2 = 35.97atm (3644 kPa) (literature value is 34.7 atm) V

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Calculation Procedure 1. Assume a basis. Assume gas is at standard conditions, that is, 1 g-mol gas at 1 atm (101.3 kPa) pressure and 0 C (273 K, or 492 R), occupying a volume of 22.4 L. 2. Compute the gas constant. Apply suitable conversion factors and obtain the gas constant in various units. Use PV = RT; that is, R = PV/T. Thus, a

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Source: Otto Frank, "Shortcuts for Distillation Design," Chemical Engineering, March 14, 1977. In the present case, hand calculation and shortcut procedures are adequate because the benzene- toluene-xylene system is close to ideal. 2. Establish the equilibrium relationship among the constituents.

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Handbook of Chemical Engineering Calculations: Chohey ...

The author Mihir M. Patel, is a Chemical Engineer with a Master's degree from USA in 1987. He is practicing chemical engineering with 33+ years in process and process safety engineering. Additionally, he is a Project Management Professional (PMP®) from PMI, USA, which gives him a unique overall project perspective in addition to process ...

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