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Note Taking Guide: Episode 303 Name Aditi Acharya Bohr's Energy Levels • Electrons in certain energy levels • Lower energy levels: closer to nucleus • Higher energy levels: further from nucleus • Ground State: electron in lowest energy level possible Excited Atom • Atom has absorbed energy • Excited state is unstable • Atom soon emits same amount of energy absorbed • energy seen as visible light Wave Description of Light Wavelength (λ): distance between corresponding points ...

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Episode 303. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. LStavar5. Terms in this set (7) Bohr's Energy Levels. 1. electrons have certain energy levels 2. low energy levels- closer to nucleus 3. high energy levels- farther from nucleus 4. ground state- electron in lowest energy level possible.

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