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## **Ph Worksheet Chemistry Answers**

This downloadable PDF worksheet is for students to practice calculating pH and pOH values from concentration values of H<sup>+</sup> and OH<sup>-</sup> ions.

Useful relationships:

$$\begin{aligned} \text{pH} &= -\log [\text{H}^+] & \text{pOH} &= -\log [\text{OH}^-] \\ K_{\text{water}} &= 1 \times 10^{-14} = [\text{H}^+] [\text{OH}^-] \\ \text{pH} + \text{pOH} &= 14. \end{aligned}$$

Review: pH

Calculations; Chemistry

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Quick Review of pH.

## **pH and pOH Practice Worksheet - ThoughtCo**

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try to answer the

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questions without referring to your textbook. If you get stuck, try asking another group for help.  
Q1.

## **pH (Worksheet) - Chemistry LibreTexts**

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the UC Davis Office of the Provost, the UC Davis Library, the California State University Affordable Learning Solutions Program, and Merlot. We also acknowledge previous National Science Foundation support under grant numbers 1246120, 1525057, and 1413739.

**pH and Titrations  
(Worksheet) -**

*Page 7/24*

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**Chemistry**  
**LibreTexts**

pH practice - Answers.

- 1) What is the pH and pOH of a  $1.2 \times 10^{-3}$  HBr solution? pH: 2.9 pOH: 11.1. 2) What is the pH and pOH of a  $2.34 \times 10^{-5}$  NaOH solution? pOH: 4.6 pH: 9.4. 3) What is the pH and pOH of a solution made by adding water to 15 grams of hydroiodic acid until the volume of the solution is 2500 mL?



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pH: 1.6 pOH: 12.4

**Acid and Base  
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Sets. CHEMISTRY is the

study of matter, its  
properties, how and  
why substances

combine or separate to  
form other substances,  
and how substances  
interact with energy.

The five main branches  
of chemistry include  
analytical chemistry,

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physical chemistry,  
organic chemistry,  
inorganic chemistry  
and biochemistry.

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This pH Review  
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chemistry activity,  
students complete 10  
short answer review  
questions about pH,  
molarity of solutions,  
acid/base indicators,  
and titration. pH  
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Worksheet for 10th -  
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Answers 1.

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Chapter Project

Worksheet 1 1.

Answers ... pH and pOH

Practice Worksheet -

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worksheet is for

students to practice

calculating pH and pOH

values from

concentration values of

H + and OH-ions.

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Chemistry  
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The sum of pH and pOH is always 14. This is because the product of proton concentration and hydroxide concentration must always equal the equilibrium constant for the ionization of water, which is equal to  $10^{-14}$ . In this question, we know that the pOH is

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equal to 2.13, allowing  
us to solve for the pH.

## Answers

### **Calculating pH and pOH - High School Chemistry**

1) -Find the pH of a  $9.8 \times 10^{-4}$  M HI solution.  $\text{pH} = -\log(9.8 \times 10^{-4}) = 3.01$ . 2) -Find the pH of a  $2.76 \times 10^{-4}$  M NaOH solution.  $-\text{pOH} = -\log(2.76 \times 10^{-4}) = 3.559$   
 $\text{pH} = 14.000 - 3.559 = 10.441$ . 3) Find the pH of a solution made by diluting 35 mL of 4.5 M

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HNO<sub>3</sub>.

## Chemistry

### **pH worksheet 2 - Everett Community College**

Objective:

Demonstrate basic knowledge of acids, base chemistry. 1. Student teams create 3D cubes about pH chemistry. 2. Student teams stack their cubes so characteristics of acid-base chemistry are aligned. 3. Student

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followup worksheet included. 4. Answer key to stack and worksheet included.

## **Ph Chemistry Worksheets & Teaching Resources | Teachers ...**

1.+Calculate the pH of a solution with  $[H^+] = 5 \times 10^{-5}M$ . 2. Calculate the pH of a solution with  $[H^+] = 1 M$ .

**Name: Date: Strong  
Acids and Bases**

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Does Chemistry Have

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How to move from

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pH Worksheet #2;

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base review -

Southwest High School.

The "It Doesn't Include  
Everything, but it's a  
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Chemistry. Lecture 31

Acids and Bases III.

**CHEMISTRY 12 pH**

**WORKSHEET -**

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Ph Scale Worksheet

Answers pH Worksheet

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Name: What is pH? pH is a measurement of how acidic or how basic a solution is. The pH scale starts at 0 and goes up to 14. Halfway between 0 and 14 is 7, which is neutral. Compounds are acidic if they have a pH lower than 7.

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