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View Answer. In the nuclear fusion process, the masses of the fusing nuclei be m_1 and m_2 and the mass of the resultant nucleus be m_3 , then (a) $m_3 > (m_1 + m_2)$ (b) $m_3 = m_1 + m_2$ (c) $m_3 = (m...$

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Chain reaction. Fusion. the emission of radiation from an unstable nucleus. a nuclear reaction where the nucleus of a large, unstable atom.... a reaction in which the products from one reaction are used to.... a nuclear reaction where two small nuclei combine, or fuse tog.... Radioactive decay.

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Chemistry First

All nuclear reactions involve some type of nuclear transmutation. Scientists induce transmutations by bombarding stable nuclei with high-energy alpha, beta, or gamma radiation. The first induced nuclear transmutation was carried out by Marie and Pierre Curie in 1897. Most induced transmutation reactions are produced in high-energy

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Radioactivity and Nuclear Chemistry. Atomic theory in the nineteenth century presumed that nuclei had fixed compositions. But in 1896, the French scientist Henri Becquerel found that a uranium compound placed near a photographic plate made an image on the plate, even if the compound was wrapped in black cloth.

CH103 - CHAPTER 3: Radioactivity and Nuclear Chemistry

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Unit 4 Nuclear Chemistry Review Study Guide. 1. Define the following terms below. a) Half-life- the length of time during which half of a given number of atoms of a radioactive nuclide decay b) Nuclear Fission- the process in which lightweight nuclei combine to form heavier, more stable nuclei c) Nuclear Fusion- process in which a very heavy nucleus splits to form medium-weight nuclei 2.

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