

Pogil The Activity Series Answer Key

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Pogil The Activity Series Answer

Model 2 — The Activity Series 10. The activity series shown in Model 2 to the right is a ranking, similar to the one you did in Question 7, of several metals. a. Are the more active metals at the top or at the bottom? Explain. ecu,' (b. Label Model 2 with "most active" and "least active" in the appropriate locations. 11, Use the activity ...

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CuCl was named copper (I) chloride because copper can have two possible charges (a +1 or a +2) so you need to use the Roman numeral (I) to show that this compound was made with copper with a +1 charge. Name the following compounds. ZnCl₂_____. CoN_____. PbS₂_____.

Chemistry POGIL Activity «Activity_»

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Pogil The Activity Series Answer Key The reactivity of the metals is exactly opposite the reactivity of

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the corresponding cation. While solid magnesium was the most reactive metal, its ion, Mg^{2+} , was the least reactive. $-2+Mg + 2e \rightarrow Mg^{2+}$ $-Zn + 2e \rightarrow Zn^{2+}+Cu +$

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