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Solutions 10 Quantum Mechanics

Tutorial 1.1 Solution The energy of a single photon is  $E = h\nu$  ( $\nu$  is the frequency and  $h$  is Planck's constant  $h = 6.626 \times 10^{-34} \text{ J s}$ ). The wavelength is related to the frequency by the speed of light in a vacuum,  $c = 2.998 \times 10^8 \text{ m s}^{-1}$ .



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