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1. How many valence

electrons does a

neutral magnesium

(Mg) atom have? 2.

What is the charge on



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a magnesium ion?  
What does magnesium  
have to do to form  
such an ion, and why  
does it tend to do so?

3. How many valence  
electrons does a single  
neutral chlorine (Cl  
atom have? 4.

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a. potassium-39 b.  
potassium-40 c.  
potassium-41 7. Write  
an equation showing-

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the-relationship  
between an atom's  
atomic number and its  
mass number. 8.  
Lithium has two  
isotopes: lithium-6 and  
lithium-7. Draw a  
diagram, like those  
shown on the  
transparency, for each  
lithium isotope. Label  
the protons, electrons,  
neutrons, and

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0 10 20 30 40 50 60 70

80 90 100 Solubility (g  
solute/100 g H<sub>2</sub>O) 90

100 80 70 60 50 40 30

20 10 0 Temperature

(oC) Solubilities as a

Function of

Temperature NaCl

KClO<sub>3</sub> KCl CaCl<sub>2</sub> Ce

2(SO<sub>4</sub>)<sub>3</sub>

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